

$$\begin{aligned}
&= k \sum_{i=1}^l \left[ n_i \ln \left( \frac{g_i + n_i}{n_i} \right) + g_i \ln \left( \frac{g_i + n_i}{g_i} \right) \right] \\
&= k \sum_{i=1}^l \left[ n_i \ln \left( \frac{g_i}{n_i} + 1 \right) + g_i \ln \left( 1 + \frac{n_i}{g_i} \right) \right]
\end{aligned}$$

**Exercise 7.** For Maxwell-Boltzmann, Fermi-Dirac and Bose-Einstein distributions, we have

$$\frac{g_i}{n_i} = e^{(\alpha + \beta E_i)} + J$$

where

$$J = \begin{cases} 0 & \text{for Maxwell-Boltzmann distribution} \\ 1 & \text{for Fermi-Dirac distribution} \\ -1 & \text{for Bose-Einstein distribution} \end{cases}$$

Find out values of the constants  $\alpha$  and  $\beta$ .

**Solution:** Since the values of  $\alpha$  and  $\beta$  do not depend on the distribution, we consider the simple case of Maxwell-Boltzmann distribution, so that

$$n_i = g_i e^{-\alpha} e^{-\beta E_i} \quad (6.30)$$

Average energy of particles in an ideal gas is

$$\bar{E} = \frac{E}{N} = \frac{\sum_{i=1}^l n_i E_i}{\sum_{i=1}^l n_i}$$

Using equation (6.30), we have

$$\bar{E} = \frac{\sum_{i=1}^l E_i g_i e^{-\alpha} e^{-\beta E_i}}{\sum_{i=1}^l g_i e^{-\alpha} e^{-\beta E_i}} = \frac{\sum_{i=1}^l E_i g_i e^{-\beta E_i}}{\sum_{i=1}^l g_i e^{-\beta E_i}}$$

When energy levels are closely packed, we can consider it as a continuum. Then,  $E_i$  is replaced by  $E$ ,  $g_i$  by  $g(E)$ , and summation over the energy states by the integration over the energy. Thus, we have

$$\bar{E} = \frac{\int E g(E) e^{-\beta E} dE}{\int g(E) e^{-\beta E} dE}$$

For the particles with no spin, we have (see appendix)

$$g(E) dE = 2\pi V \left( \frac{2m}{h^2} \right)^{3/2} E^{1/2} dE$$

- (ii) Fermi-Dirac distribution
- (iii) Bose-Einstein distribution
- (iv) Condition for application of Maxwell-Boltzmann distribution

Using Stirling formula, we have

$$\begin{aligned}\ln W &= N \ln N - N + \sum_{i=1}^l (n_i \ln g_i - n_i \ln n_i + n_i) \\ &= N \ln N - \sum_{i=1}^l n_i \ln \left( \frac{n_i}{g_i} \right)\end{aligned}\quad (7.16)$$

and for the maximum value of  $W$ , we have

$$n_i = g_i \frac{N}{V} \left( \frac{h^2}{2\pi m k T} \right)^{3/2} e^{-E/kT} \quad (7.17)$$

Using equation (7.17) in (7.16), we have

$$\begin{aligned}\ln W &= N \ln N - \sum_{i=1}^l n_i \left( \ln N + \ln \left[ \frac{1}{V} \left( \frac{h^2}{2\pi m k T} \right)^{3/2} \right] - \frac{E}{kT} \right) \\ &= N \ln N - N \ln N - N \ln \left[ \frac{1}{V} \left( \frac{h^2}{2\pi m k T} \right)^{3/2} \right] + \frac{EN}{kT} \\ &= N \ln \left[ V \left( \frac{2\pi m k T}{h^2} \right)^{3/2} \right] + \frac{EN}{kT}\end{aligned}$$

Using  $E = 3kT/2$ , we have

$$\ln W = N \ln \left[ V \left( \frac{2\pi m k T}{h^2} \right)^{3/2} \right] + \frac{3}{2} N \quad (7.18)$$

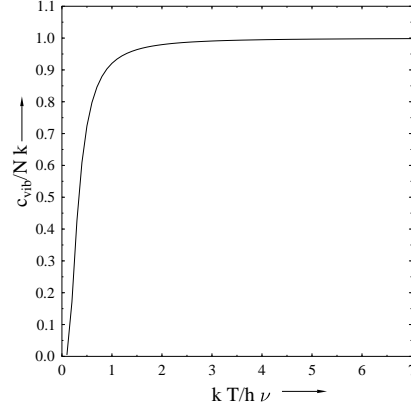
Thus, the entropy (equation 7.18) is

$$S = kN \ln \left[ V \left( \frac{2\pi m k T}{h^2} \right)^{3/2} \right] + \frac{3}{2} kN \quad (7.19)$$

### 3.4 Gibbs paradox

In order to check the additive property of entropy of the gas, let us increase each of the volume  $V$  and number of particles  $N$  of the gas by a common factor  $p$ , then the entropy  $S'$  of the new system (using equation 7.19) is

$$\begin{aligned}S' &= kpN \ln \left[ pV \left( \frac{2\pi m k T}{h^2} \right)^{3/2} \right] + \frac{3}{2} kpN \\ &= p \left( kN \ln \left[ V \left( \frac{2\pi m k T}{h^2} \right)^{3/2} \right] + \frac{3}{2} kN \right) + pNk \ln(p)\end{aligned}\quad (7.20)$$

Figure 3: Variation of  $(c_{\text{vib}}/Nk)$  versus  $kT/h\nu$ .

rotational level with the quantum number  $J$  is  $(2J + 1)$ . The rotational partition function is

$$Z_{\text{rot}} = \sum_{J=0}^{\infty} (2J + 1) \exp[-\sigma_r J(J + 1)]$$

where  $\sigma_r = B/kT$ .

(i) At low temperatures ( $kT \ll B$ , i.e.,  $\sigma_r \gg 1$ ), we have

$$Z_{\text{rot}} = 1 + \sum_{J=1}^{\infty} (2J + 1) \exp[-\sigma_r J(J + 1)]$$

(a) The Helmholtz free energy is

$$\begin{aligned} F_{\text{rot}} &= -NkT \ln Z_{\text{rot}} = -NkT \ln[1 + \sum_{J=1}^{\infty} (2J + 1) \exp\{-\sigma_r J(J + 1)\}] \\ &= -NkT \sum_{J=1}^{\infty} (2J + 1) \exp[-\sigma_r J(J + 1)] \end{aligned}$$

where we used the expansion

$$\ln(1 + x) = x - \frac{x^2}{2} + \frac{x^3}{3} \dots$$

and neglected the quadratic and higher order terms.

(b) Entropy  $S_{\text{rot}}$  of the gas is

$$S_{\text{rot}} = -\frac{\partial F_{\text{rot}}}{\partial T} = Nk \sum_{J=1}^{\infty} (2J + 1) \exp[-\sigma_r J(J + 1)]$$